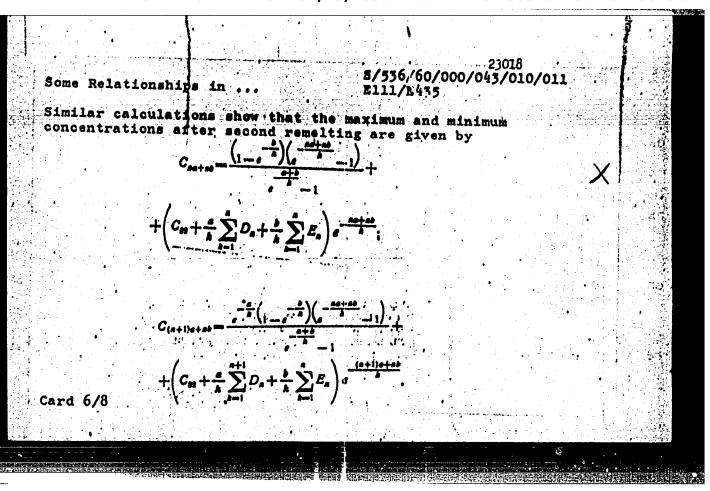


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Some Relationships in ..

S/536/60/000/043/010/011 E111/E435

where Co2 is the concentration in the initial liquid

$$E_{n} = C_{01} - e^{\frac{\alpha}{n}} + e^{\frac{-\alpha + \delta}{n}} - e^{\frac{(n-1)\alpha + (n-1)\delta}{n}} - e^{\frac{n\alpha + (n-1)\delta}{n}}$$

$$D_{n} = C_{01} - e^{\frac{\alpha}{n}} + e^{\frac{-\alpha + \delta}{n}} - e^{\frac{-\alpha + (n-1)\delta}{n}} + e^{\frac{n\alpha + n\delta}{n}}$$

The equations deduced were verified by zone melting with rapid movement of the liquid zone, this being the easiest to carry out. The rate of movement of the furnace was 5 mm/min, the length of the liquid zone being 60 mm. A composite billet (corresponding to the composite electrode) was made up from plates of pure lead and pure zinc to give an average composition of 90% Pb, 10% Sn. Specimens made from electrodes with various ratios of the volume of the single lead-tin portion to that of the liquid zone were used. After zone melting, the specimens were cut into plates 2.5 mm long and analysed. The experimental and theoretical curves of tin content as functions of length along specimen agree satisfactorily. Relative longitudinal fluctuations of composition were calculated Card 7/8

Some Relationships in ... S/536/60/000/043/010/011 El11/E435

and were also found to support the theoretical equations: the smaller the volume of the portion relative to that of the liquid bath the greater the uniformity of the billet. The authors emphasize that although the ideas of this paper have been developed for zone melting they can be applied to billets obtained from a consumable electrode. There are 7 figures and 3 Soviet.references.

18.1000

8/536/60/000/043/011/011 E111/E435

AUTHORS:

Petrov, D.A., Doctor of Chemical Sciences, Professor and Kolachev, B.A., Candidate of Technical Sciences

TITLE:

Non-Equilibrium Crystallization of Ternary Alloys

PERIODICAL: Moscow. Aviatsionnyy tekhnologicheskiy institut. Trudy. No.43. 1960. pp.117-129. Termicheskaya obrabotka i svoystva stali i legkikh splavov

TEXT: D.A.Petrov has shown (ZhFKh, 1947, T.XXI, No .12) that alloy crystallization can be considered as two processes occurring in parallel: separation of crystals of the solid phase from the liquid and change in the composition of crystals formed at a higher temperature through reaction with the liquid at a lower temperature. The authors now consider the crystallization of an alloy with two alloying components, with no diffusion in the solid state and a continuous series of solid solutions. For equilibrium conditions the changes in liquid and solid compositions as crystallization proceeds can be found from phase diagrams with the aid of Konovalov's rule. For non-equilibrium conditions crystallization is not completed at the temperature corresponding to the intersection of the alloy ordinate with the solidus surface. Card 1/6

Non-Equilibrium Crystallization ... 5/536/60/000/043/011/011 E111/E435

Crystallization in the assumed system then ends at the fusion temperature of the lowest-melting component. In non-equilibrium crystallization of alloys belonging to a system with four-phase eutectic transformation the lines showing changes of liquid- and solid-phase composition changes will also be displaced from the equilibrium lines depending on the overall conode position in the primary-crystallization region. For the conditions specified the crystallization of any alloy of the ternary system is completed with the crystallization of the ternary eutectic. Non-equilibrium crystallization of ternary-system alloys with a peritectic fourphase transformation ends with the solidification of the binary $\beta + \gamma$ eutectic. For the experimental verification of their ideas the authors chose the method of drawing would phase from the melt, since this largely satisfies the condition specified in the theoretical treatment. Transformation of the equations deduced gives the distribution of components along the drawn specimen, but through lack of data the authors had to confine themselves to a qualitative verification. The systems Al-Cu-Si and Al-Cu-Mn were chosen, for which phase diagrams can be constructed from published data (H.W.Phillips, J.Inst. of Metals, 1953, T.82,p.9-15:

Non-Equilibrium Crystallization ... E111/E435

H.W.L.Phillips, W.Day, J.Inst. of Metals, 1947, 74, p.33-47). The test compositions were: Al + 4% Cu + 3% Si; A1 + 8% Cu + 1.0% Si; A1 + 4% Cu + 0.6% Mn; A1 + 2.25% Cu + 1% Mn; A1 + 0.5% Cu + 1.3% Mn. Aluminium (99.98% A1, 0.02% (Fe + Si)) electrolytic copper and manganese, and silicon (0,25% Fe, 0,20% Al) were used for preparing alloys; copper, silicon and manganese being introduced as alloys. Specimens were drawn at 0.07 mm per minute in the apparatus previously described by Petrov and Bukhanova (ZhFKh, 1953, T.27, No .1). After microstructus After microstructural examination, samples of the solid were taken for chemical analysis; liquid-phase compositions were calculated. Fig.7 shows changes in the copper and silicon contents for the solid and liquid phases with respect to relative length (continuous lines relate to liquid and broken lines to solid phases, respectively); the corresponding curves for copper and manganese distribution are shown in Fig.9. These results and microstructure-examination show that not all the range of composition expected from the theoretical treatment is found. This is due to the fact that at low concentrations of the alloying components the range of the binary Card 3/6

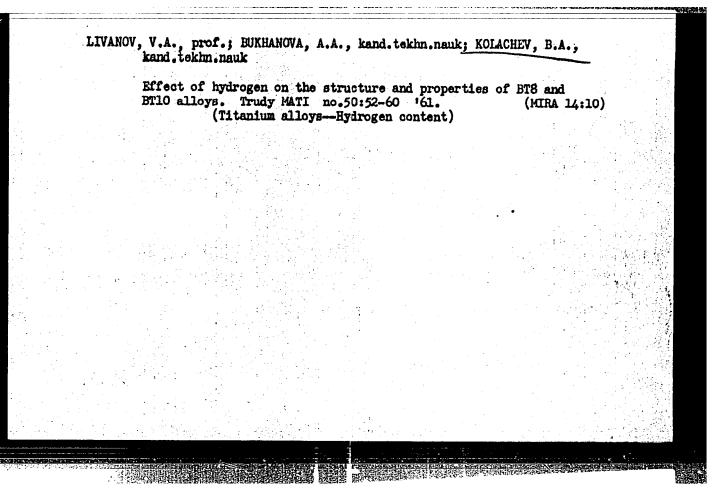
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Non-Equilibrium Crystallisation ...

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eutactic is very small. For example, in the alloy obtained from Al + 4.5% Cu + 0.5% Si, 0.91 of the specimen will consist only of α-solid solution (of variable composition). Thus the binary and ternary eutactics crystallize at the last moment, when drawing conditions are already disturbed and complete replacement of one structural component by another does not occur. Nevertheless, the general change of composition and microstructure confirms the theoretical treatment both for drawing and for non-equilibrium crystallization in general. There are 10 figures and 3 references: 1 Soviet-bloc and 2 non-Soviet-bloc. The two references to English language publications read as follows: H.W.L.Phillips, J.Inst. of Metals, 1953, T.82, p.9-15; H.W.L.Phillips, W.Day, J.Inst. of Metals, 1947, 74, p.33-47.

Card 4/6



18.1285

309 25 S/536/61/000/050/007/017 D217/D304

AUTHORS:

100

Livanov, V.A., Professor, Bukhanova, A.A., and Kolachev, B.A., Candidates of Technical Sciences

TITLE:

Influence of grain size on the hydrogen embrittlement of

titanium and its alloys

SOURCE:

Moscow. Aviatsionnyy tekhnologicheskiy institut. Trudy,

no. 50, 1961, Voprosy metallovedeniya, 61-70

TEXT: The main purpose of this paper was to investigate the influence of hydrogen on the mechanical properties of fine grained and coarse grained titanium and its alloys. Specimens of commercially pure titanium were made from forged rods and annealed in vacuo at 700 C, 900 C and 1100 C. Annealing at 700 C results in a fine-grained structure; at 900 C, medium-sized grains form, whilst at 1100 C the structure becomes coarse-grained. After vacuum annesling, the specimens were furnacecooled. Various quantities of hydrogen were then introduced into them at the same temperatures at which vacuum annealing had been carried out.

Card 1/2

Influence of grain ...

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Soaking time before and after saturation with hydrogen at the above temperatures was one hour in each case. The subsequent cooling was carried out in the furnace. The dependence of the mechanical properties of Ti on hydrogen content was studied after vacuum annealing at 1100°C and saturation with hydrogen at 1100°C and 900°C, and the microstructure of Ti saturated with hydrogen at various temperatures was compared with that of commercially pure Ti after vacuum annealing at the same temperacy to hydrogen embrittlement than fine-grained metal had a greater tendency to hydrogen embrittlement than fine-grained material; this is due to differences in the nature of the hydride precipitates. In the fine-grained material, Ti hydrides separate along the grain boundaries in the form of compact, often formless, precipitates. In the coarse-grained material, Ti hydrides precipitate in the form of very fine platelets. This fine precipitate causes high stress concentrations and premature destruction of the metal. There are 8 figures.

Card 2/2

18.1285

30926 s/536/61/000/050/008/017 D217/D304

AUTHORS:

Livanov, V.A., Professor, Buchanova, A.A. and Kolachev,

B.A. Candidates of Technical Sciences

TITLE:

Influence of hydrogen on the thermal stability of the

alloy **B7** 3-1 (VT3-1)

SOURCE:

Moscow. Aviatsionnyy tekhnologicheskiy institut. Trudy,

no. 50, 1961, Voprosy metallovedeniya, 71-81

TEXT: Specimens of alloy VT-3-1 were annealed in vacuo at 900°C for 6 hours. Mechanical tests were carried out after various isothermal annealing treatments on specimens of various hydrogen contents, at three different deformation rates: (a) 40 mm/minute (b) 4 mm/minute and (c)

minute. After vacuum annealing; the microstructure of the VT3-1 nsists of a supersaturated & -solid solution containing a small f of the \$\beta\$-phase. The structure of the alloy on being saturated drogen immediately after vacuum annealing remains essentially unci. Isothermal annealing at 450°C for 48 hours leads to a

/3

30926 S/536/61/000/050/008/017 D217/D304

Influence of hydrogen ...

decomposition of the supersaturated solution and to the precipitation of the TiCr phase, by sutectoid decomposition of β . The higher the temperature, the greater the rate of eutectoid decomposition. It is found that isothermal annealing leads to embrittlement of the VT3-1 alloy which is the more pronounced the higher the annealing temperature. Embrittlement is noticeable only after isothermal annealing at 550°C for over 100 hours. Hydrogen lowers the thermal stability of the alloy. The brittleness of an alloy containing more than 0.03% hydrogen manifests itself even after annealing at 350°C for 100 hours. The decrease in thermal stability of a VT3-1 alloy containing hydrogen is due to the fact that the latter accelerates decomposition of the A phase and of the supersaturated & esolid solution. Besides, in the presence of hye drogen, Ti hydride or any other phase containing hydrogen, precipitation of phases other than TiCr also occurs. Hydrogen lowers the thermall stability of the alloy VT321 to a lesser degree than that of the alloy VT3, since the Buphase in the former is more stable than in the latter. There are 7 figures and 3 references: 1 Sovietabloc and 2 non-Soviet-bloc. The reference to the English-language publication reads

Card 2/3

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18 1285

30927 \$/536/61/000/050/009/017 D217/D304

AUTHORS:

Livanov, V.A., Professor, Buchanova, A.A., and Kolachev,

B.A., Candidates of Technical Sciences

TITLE:

Influence of oxygen and hydrogen on the structure and pro-

perties of titanium

SOURCE:

Moscow. Aviatsionnyy tekhnologicheskiy institut. Trudy,

no. 50, 1961, Voprosy metallovedeniya, 82-92

TEXT: The combined influence of oxygen and hydrogen on the mechanical properties and structure of Russian commercially pure titanium was investigated. Ingots were melted in a laboratory arc furnace, using a soluble segmented electrode. The electrodes were compacted from sponge containing the following impurities: 0.1% Fe, 0.05% Si, 0.05% Mg, 0.05% Cl, Cl. SO₂, 0.01% H₂, 0.03% N₂ and 0.03% Ni. Oxygen was added to each portion of the electrode in the form of calculated quantities of TiO₂. By this method, ingots with the following supplementarily added oxygen Card 1/4

X

30927 s/536/61/000/050/009/017 D217/D304

Influence of oxygen ***

contents were made: 0, 0.06, 0.1, 0.2, 0.3, 0.5 and 1.0 wt.%. After the first remelting, the ingots were ground and forged. The forged billets were then used as electrodes for the second remelting process. The inc gots obtained by double remelting were forged into rods of 12 x 12 mm cross section at 980-1000 C. After hot forging, the rods were cooled in air and cut into sections for specimens for mechanical testing. The mechanical test specimens were vacuum annealed at 900°C for 6 hours, after which they were furnace couled. They were then saturated with hydrogen to various concentrations. The hydrogen content of the species mens was determined from the change in hydrogen pressure in a system of known wolume, and from the gain in weight of the specimens. After being saturated with hydrogen, the specimens were furnaceccooled. Their men chanical properties were determined at room temperature. After testing, the microstructure of undeformed portions of the specimens was studied. The oxygen content of the alloys was determined by the equilibrium pressure of hydrogen introduced into it. It was found that the joint presence of oxygen and hydrogen in Ti greatly affects the structure and properties

Card 2/4

30927 S/536/61/000/050/009/017 D217/D304

Influence of oxygen ...

of the latter. At low contents of these impurities (up to 0.3 wt.% $^{0}2$ and up to 0.03 wt. % H2) hydrogen does not exert a noticeable influence on the strength of Ti, but seriously reduces the plasticity characters istics, particularly the impact resistance. At high oxygen contents, hydrogen sharply decreases the strength and plasticity of Ti. In amounts not exceeding 0.5-0.7 wt.%, oxygen sharply increases the U.T.S. and yield strength. 0.01 wt.% oxygen increases the U.T.S. and yield strength of Ti by 1.3 kg/sm2. In the joint presence of H2 and O2 in Ti and its alloys, a Ti hydride precipitate appears. The latter is characterized by a greater degree of dispersion at greater oxygen contents of Ti. Oxygen does not appear to have a great influence on the solubility of hydrogen in Astitanium at room temperature. There are 12 figures and 6 non-Soviet-bloc references. The 4 most recent references to the English-language publications read as follows: T.S. Liu, M.A. Steinberg, Transaction of the American Society for Metals, 1957, 50, Preprint no. 34; G.A. Lenning, C.M. Craighead, R.I. Jaffee, J. of Metals, 1954, v.6, p. 367; G. Weinig, J. of Metals, 1957, v. 9, no. 10; G.A. Lenning, J.W. Card 3/4

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30928 S/536/61/000/050/010/017 D217/D304

AUTHORS:

Livanov, V.A., Professor, Buchanova, A.A., and Kolachev,

B.A., Candidates of Technical Sciences

TITLE:

Hydrogen embrittlement of titanium-aluminum alloys

SOURCE:

Moscow. Aviatsionnyy tekhnologicheskiy institut. Trudy,

no. 50, 1961, Voprosy metallovedeniya, 93-102

TEXT: The purpose of the present work was to investigate the influence of Al, one of the main alloying elements of many industrial Ti alloys, on the hydrogen embrittlement of Ti. To study the influence of Al on the mechanical properties and structure of Russian technically pure Ti in the presence of hydrogen, Ti-Al alloy ingots were made in a laboratory arc furnace, using soluble segmented electrodes. The electrodes were compacted from TiO sponge. Ingots containing 0, 3, 4, 7.5 and 10% Al were made. After the first remelting, the ingots were ground and forged. The forged billets were then used as electrodes for the second melting. The ingots obtained after repeated remelting were forged into rods of Card 1/2

30928 \$/536/61/006/050/010/017 D217/D304

Hydrogen embrittlement ...

14 x 14 mm cross section at 1050°C. After hot forging, the rods were cooled in air and cut into sections for spegimens for mechanical testing. The specimens were annealed in vacuo at 900°C for 6 hours, after which they were furnace cooled. They were then saturated with hydrogen to various concentrations and again furnace-cooled. Mechanical testing of the hydrogen-saturated specimens was carried out at room temperature. The microstructure was studied, using the undeformed portions of impact text pieces. It was found that Al reduces the tendency of Ti to hydrogen embrittlement; this is due to the increased solubility of hydrogen in the d -solid solution and to the retardation of the diffusion of hydrogen in Ti in the presence of Al. The maximum permissible hydrogen content of a Ti alloy containing 5% Al (VT5) is approximately 0.03%, i.e. twice that permissible for commercially pure Ti. There are 10 figures and 3 non-Soviet-bloc references. The references to the English-language publications read as follows: H.R. Ogden, D.I. Maykath, W.L. Finlay, R.I. Jaffee, J. of Metals, 1953, v. 5, no. 2, II, 267; G.A. Lenning, J.W. Spretnak, R.I. Jaffee, J. of Metals, 1956, vo. 8, no. 10, II.

Card 2/2

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s/536/61/000/050/011/017 D217/D304

AUTHORS:

Livanov, V.A., Professor, Musatov, M.I., Engineer, and

Kolachev, B.A., Candidate of Technical Sciences

TITLE:

Distribution of alloying elements in a titanium ingot

produced by melting with the soluble segmented electrode

SOURCE:

Moscow. Aviatsionnyy tekhnologicheskiy institut. Trudy,

no. 50, 1961, Vopromy metallovedeniya, 103-116

TEXT: The main disadvantage of Ti ingots produced by arc melting with a. soluble electrode is their inhomogeneity with respect to chemical composition. By mathematical calculations, it was found that a much more uniform distribution of alloy elements along the length and cross section of ingots could be obtained by melting with a soluble segmented electrode. The relative inhomogeneity of distribution of the alloying components in such an ingot depends essentially on the ratio between the volume of the active portion of the electrode and that of the molten bath. If the ratio between the molten bath volume and the electrode volume is

Card 1/2

KOLACHEV B. A.

PHASE I BOOK EXPLOITATION

SOV/6171

Livanov, Vladimir Aleksandrovich, Anna Arkhipovna Bukhanova, and Boris Aleksandrovich Kolachev

Vodorod v titane (Hydrogen in Titanium). Moscow, Metallurgizdat, 1962. 244 p. Errata slip inserted. 2900 copies printed.

Ed.: L. P. Luzhnikov; Ed. of Publishing House: M. S. Arkhangel-skaya; Tech. Ed.: L. V. Dobuzhinskaya.

PURPOSE: This book is intended for scientific workers, engineers, and technicians at plants and scientific research institutes engaged in the production, treatment, and application of titanium and its alloys. It may also be useful to aspirants and senior students at schools of higher technical education, who specialize in physical metallurgy, technology of heat treatment, casting, forming, and welding of metals. It may likewise be of interest to design engineers.

Card 1/62

PETROV, D.A., prof., red.; KOLACHEV, B.A., kand. tekhn. mauk
[translator]; L'VOVA, N.M., red.; PRIDAMTSEVA, S.V., tekhn. red.

[New data on the production of single crystals of semieonductors] Novoe v poluchenii monokristallov poluprovodnikov; sbornik statei. Moskva, Izd-vo inostr. litry, 1962. 259 p. Translated from the Englich.

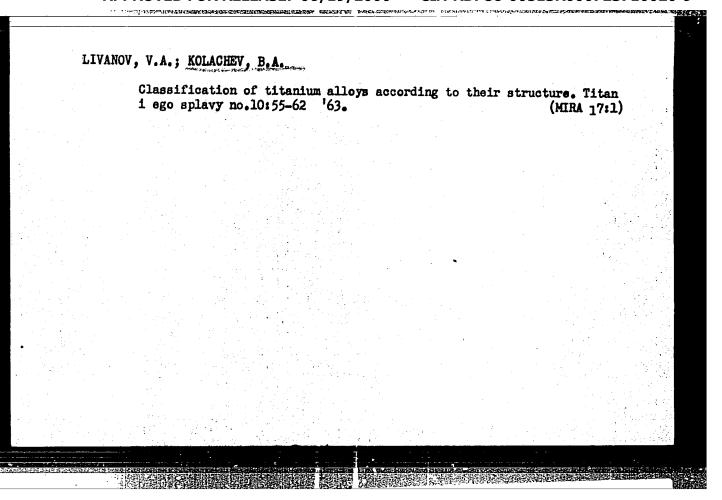
(Crystals—Growth) (Semiconductors)

(Crystals—Growth) (Semiconductors)

TROSTYANSKAYA, Ye.B.; SHISHKIN, V.A.; SIL'VESTROVICH, S.I.; PANTELEYEV,
A.S.; FOLUBOYARINOV, D.N.; BALKEVHICH, V.L.; NATANSON, A.K.;
KOLACHEV, B.A.; PETROV, D.A.; GOL'DBERG, M.M.; SHAROV, M.Ya.,
inzh., retsenzent; KITAYGORDISKIY, I.I., doktor tekhn. nauk,
prof., retsenzent; ILYANOV, V.A., kand. tekhn. nauk, prof.,
retsenzent; TROSTYANSKAYA, Ye.B., red.; BABUSHKINA, S., ved.
red.; TITSKAYA, B.F., ved. red.; VORONOVA, V.V., tekhn. red.

[New kinds of materials in engineering and industry]Novye materialy v tekhnike. Pod red. Trostianskoi E.B., Kolachwa,
B.A., Sil'vestrovicha S.I. Moskva, Gostoptekhizdat, 1962.
656 p. (Materials)

(Materials)



5/2596/63/000/010/0507/0316

ACCESSION NR: AT 4007054

AUTHOR: Livanov, V. A.; Bukhanova, A. A.; Kolachev, B. A.; Gusel'nikov, N. Ya.

TITLE: Hydrogen embrittlement of titanium alloys

SOURCE: AN SSSR. Institut metallurgii. Titan i yego splavy*, no. 10, 1963. Issledovaniya titanovy*kh splavov, 307-316

TOPIC TAGS: titanium alloy, VT-3-1 titanium alloy, titanium alloy embrittlement, titanium alloy hydrogen embrittlement, hydrogen embrittlement, VT-3-1 alloy embrittlement, VT-4 titanium alloy, VT-5 titanium alloy, VT-10 titanium alloy

ABSTRACT: It has been stated that hydrogen exerts a detrimental effect on the mechanical properties of titanium and its alloys. Introduction of small quantities of hydrogen into titanium and its alpha alloys drastically reduces their impact strength. Unlike alpha alloys, the alpha-beta alloys do not exhibit hydrogen embrittlement during impact ductility alloys, the alpha-beta alloys do not exhibit hydrogen embrittlement of the alpha-tests, but only in tests at small strain velocities. Hydrogen embrittlement of the alpha-beta alloy VT-3-1 and of the alpha alloys VT-4, VT-5, and VT-10 was studied by the authors at various hydrogen concentrations (0.002 — 0.05%) and strain velocities (0.1 — authors at various hydrogen concentrations (0.002 — 0.05%) and strain velocities (4 mm/min), and after different heat and natural aging treatments. The mechanical

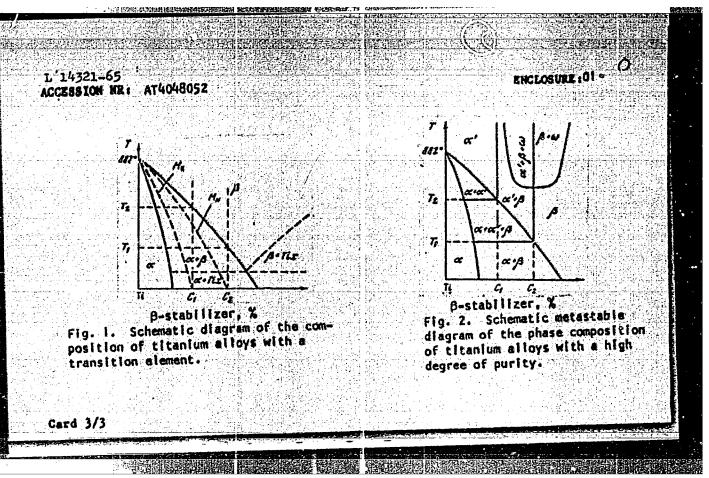
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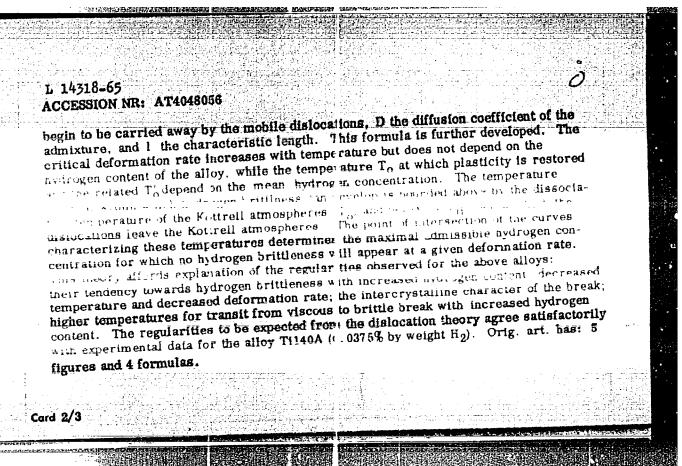
properties measured in the tests conducted by the authors are the ultimate tensile strength, yield strength, specific elongation, and contraction of cross-sectional area of the test specimen. It was concluded that: (1) Alpha-beta alloys exhibit hydrogen embrittlement at low strain velocities and this embrittlement is assisted by low temperature and by the presence of notches. (2) A certain minimum hydrogen content is required for the development of alpha-beta alloy embrittlement. After standard heat treatment alloy VT-3-1 exhibits hydrogen embrittlement at a hydrogen content exceeding 0.03%; after quenching, however, alloy VT-3-1 shows hydrogen embrittlement at 0.01%. This embrittlement is accompanied by a reduction of plasticity and an increase of tensile strength. The decrease of plasticity appears, not immediately after quenching, but in the process of natural aging after quenching. (3) Titanium-base alpha alloys VT-4, VT-5, and VT-10 like the alpha-beta alloys, exhibit hydrogen embrittlement at low strain velocities. This can be explained by a regrouping of hydrogen under the influence of stresses. Consequently, it is necessary to revise the existing mechanism explaining the brittle fracture of alpha-beta alloys caused by hydrogen. It has been suggested that hydrogen embrittlement of alpha-beta alloys is caused by processes developing in both alpha and beta phases: hydrogen diffuses toward microdefects or grain boundaries where a formation of micrevolumes enriched with hydrogen takes place; at hydrogen concentra-

1, 14321-65 EMT(m)/EMP(b)/EMP(t) ASD(n)-3/IJP(c) JD/MLK \$/0000/64/000/000/0054/0057 ACCESSION NR: AT4048052 AUTHOR: Kolachev, B. A.; Livano, V. A. TITLE: The relationship of the structures a laing during quenching of titanium alloys to structural equilibrium curves SOURCE: Soveshchaniye po metalfurgit, metal lovedentyu i primenentyu titana i yago splavov. 5th, Moscow, 1963. Metallovedenly titana (Metallography of titanium); trudy* soveshchaniya. Moscow, Izd-vo Nauka, 1964, 54-5/ TOPIC TAGS: alloy structure, alloy phase transformation, titanium alloy, quenching, phase diagram, martensite ABSTRACT: Recently, many diagrams of the metastable phase composition of titanium alloys have been published, illustrating the structure of the alloys by showing phase composition as a function of temperature, but always after quenching. In principle, these diagrams have little to do with phase equilibrium curves unless the characteristics of the Baphase are already known and can be applied. In their consideration of alloys of titanium with a 4-stabilizer, called a transition element, the authors neglect the formation of eutectoid compounds, since with titanium, they are formed only exceedingly clowly and have a negligible influence on the system after quenching. They also postulate that the concentration of Card

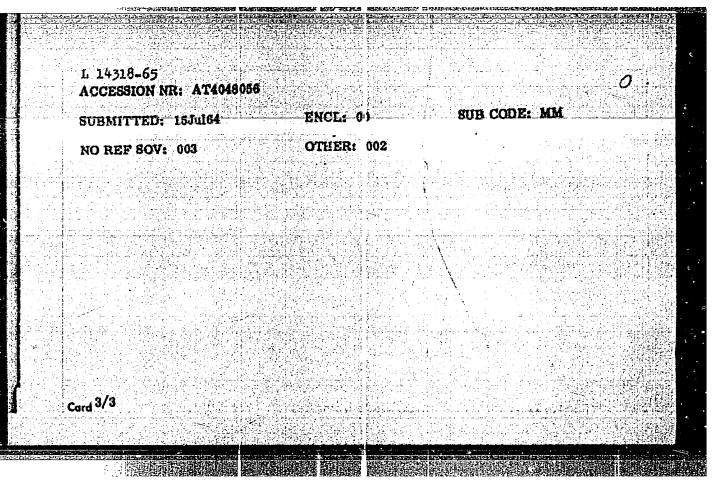
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ACCESSION NR: AT4048052		\boldsymbol{c}
Fig. 1 of the Enclosure). transformation occurs. The concentration of the β-stab no martensitic transformation.	below the critical concentration reaction goes to completion (repr A line M _K is constructed below when the pick two temperatures. The filizer in the β-phase is above crion; above T ₂ no β-phase remains. The selected, C ₁ and C ₂ ; Below C ₁ there is enough β-stabilizer to constrain of the suggisted metastable and server of the suggisted metastables.	aich no martensitic and To: Below To the itical and there is Lastly, two concen- only the martensitic
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AUTHOR: K	olachev. B.A., Livano	v, V.A., Bukhano	va. A.A.	B	
TITLE: Dis	ocation theory of the h	ydroger brit lenes	of titanium alloys		200
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AUTHOR: Livanov, V.A., Bukhemova, A. A.; Kola hev. B. A.; Neverova-Skobeleva, 29	Mr.
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at make the properties of titanium and OT4-1 alloy	
TITLE: Effect of hydrogen on the mechanical properties of titanium and OT4-1 alloy	
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SOURCE: IVUZ. Taveinaya metallurgiya, no. 4, 19 M, 124-129	
TOPIC TAGS: titanium, titanium alloy, titanium mechanical property, titanium alloy	
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ABSTRACT: The aim of this work was to study the introduced or try to stablish the properties of OT4-1 alloy, particularly on the impact strength, and to establish the properties of OT4-1 alloy, particularly on the impact strength, and to establish the	
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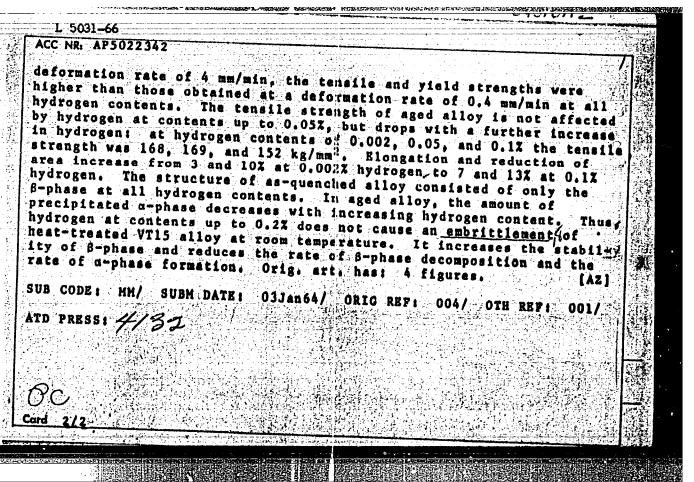
153161-65 EWT(m)/EWA(d)/:/EWP(t)/EWP(z)/EWP(z)/EWA(e) IJP(e) MJW/JD UR/0129/65/000/005/0009/0015 ACCESSION NR: AP5013151	
AUTHOP: Kolachev, B. A.; Livanov, V. A.; Bukhanova, A. A.; Gusel'nikov, N. Ya.	
The effect of hydrogen on the mechanical properties of quenched it alloys	
SOURCE: Metallovedeniye i termicheskaya obrebotka metallov, no. 5, 1965, 9-15	13
fitanium allow metal mechanica property	
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SSION NR: AP5013151 s. Tests were run on V-8 alloy at high and low speeds of deformation. The shirld properties did not change signific mtly, even at high contents of hydrometric did not change signific mtly, even at high contents of hydrometric did not change in the quenched condition shows a fincreasing hydrogen contents. In general, increasing the H level reformation changes in the structures after prolonge increased amount of martensitic chase in the sheaters. We changes the structures after prolonge income transfer of the structures after prolonge income transfer
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AUTHOR FROM ALLIVATION, V. A.; Bukha sova, A. A.; Gusel'nikov, N. Ya. 13	
TITLE: Effect of cooling rate on the tendency of titalium array	
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TOPIC TAGS: titanium alloy, hydrogen brittleriss, tensile strong, the effect of h	y -
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view, it was shown that these alloys, the cross-bars of the tensile impact	L
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L 5031-66 ENT(1)/ENT(m)/ENA(d)/ENP(t)/IMP(z)/ENP(b) IJP(c) MJM/JD SOURCE COLE: UR/0149/65/000/003/0131/0135 ACC NR: AP5022342 AUTHOR: Kolachev, B. A.; Livanov, V. A.; Bukhanova, A. A. : Gusel'nikov, N. 18. 14,55 44,55 ORG: Hoscow Aviation Technological Institute (Hoskovskiy aviatsionnyy tekhnologicheskiy Institut) 44.55 TITLE: Effect of hydrogen on the structure and properties of VT15 alloy SOURCE: IVUZ. Tsvetnaya metallurgiya, no. 3, 1965, 131-135 TOPIC TAGS: alloy, titanium alloy, aluminum containing alloy, molybdenum containing alloy, chromium containing alloy, hydrogen containing alloy, alloy structure, alloy property/VT15 alloy ABSTRACT: The effect of hydrogen on the structure and properties of VT15 B-aluminum alloy (3.7% Al, 7.35% Ho, 10.6% Cr, 0.11% Fe, 0.04% Si, 0.03%C, and 0.12% O2) has been investigated. Forged bars 14 x 14 x 70 mm of twice vacuum-arc melted alloy were vacuum annealed at 900C for 6 hr. Impregnated with hydrogen, annealed at 780C for 1 hr, and water quenched. Some bars after quenching were aged at 480C for up to 24 hr. It was found that the tensile and yield strengths of as-quenched alloy increased somewhat as the hydrogen content increased from 0.1 to 0.2%; the elongation and reduction of area dropped, however, the latter from 65.8% at 0.002% hydrogen to 53.4% at 0.2% hydrogen. UDC: 669,295



KCLAGHEV, B.A.; LIVANOV, V.A.; BUKHANOVA, A.A.; GUSYL'NIKOV, N.Ya.

**Greet of hydrogen on the mechanical properties of hardened titenium alloys. Metalloved. i term. obr. met. no.519-15 My *65.

(MIRA 18:7)

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KOLACHEV, P.A.; LIVANOV, V.A.; BUKHANOVA, A.A.; GUSEL'NIKOV, N.Ya.

Effect of hydrogen on the structure and properties of the
VI15 alloy. Izv. vys. ucheb. zav.; tavet. met. 8 no.3:131135 '65. (MIRA 18:9)

1. Moskovskiy aviatsionayy tekhnologicheskiy institut, kafedra
metallovedeniya i termicheskoy obrabotki.

LIVANOV, V.A.; BUKHANOVA, A.A.; KOLACHEY. B.A.; NEVEROVA-SKOBELEVA, N.P.; SLAVINA, I.I.; SHEYNIN, E. Ye.; SHCHERBINA, L.V.

Effect of hydrogen on the mechanical properties of titanium and the O14-1 alloy. Izv. vys. ucheb. zav.; tsvet. met. 7 no. 4: 124-129 164 (MIRA 19:1)

1. Moskovskiy aviatsionnyy tekhnologicheskiy institut, kafedra metallovedeniya i tekhnologii termicheskoy obrabotki.

KOLACHEV, B.A.; LIVANCV, V.A.; BUEHANOVA, A.A.; GUSFL'NIKOV, N.YA.

Effect of the rate of cooling on the tendency of A-titenium alloys toward hydrogen brittleness. Izv.vys.ucheb.zav.; tsvet.met. 8 no.2:131-135 *65. (MIRA 19:1)

1. Kafedra metallovedeniya i termicheskoy obrahotki Moskovskogo aviatsionnogo tekhnologicheskogo instituta. Submitted January 3, 1964.

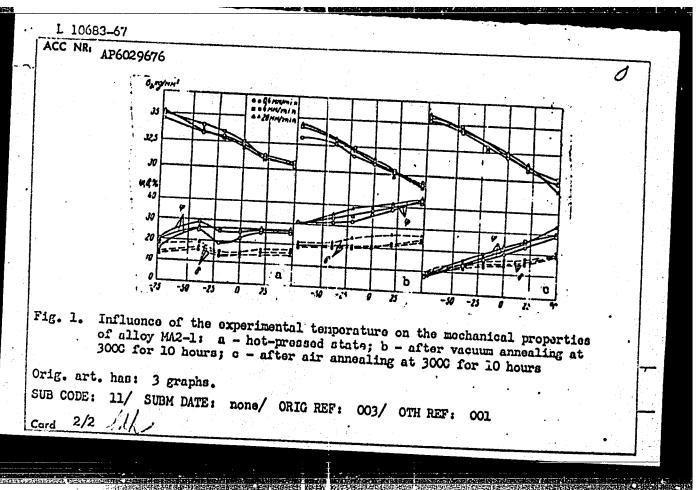
L 38552-66 EWI(m)/EWP(k)/T/EWP(w)/EWP(t)/ETI ACC NK: AT6012393 AUTHORS: Kolachev, B. A.; Livanov, V. A.; Gusel'nikov, N. Ya.; Bukhanova, A. ORG: none 17 0+1 TITLE: On certain general principles of the occurrence of hydrogen brittleness in alloys VT3-1, and VT15 SOURCE: Soveshchanive po metallokhimii, metallovedeniyu i primeneniyu titana i yego splavov, 6th. Novyye issledovaniya titanovykh splavov (New research on titanium alloys); trudy soveshchaniya. Moscow, Isd-vo Nauka, 1965, 212-220 TOPIC TAGS: A crack propagation, titanium containing alloy, alloy, martenaite alloy, material deformation, hydrogen embrittlement / VT3-1 martensite alloy, VT15 alloy ABSTRACT: A review is made of certain principles of hydrogen brittleness in alloys VT3-1 and VT15. The brittleness of (CX + B)-titanium alloy VT3-1 is more intense at temperatures below room temperature and at low rates of deformation. B-titanium alloys at lower-than-room temperatures also tend toward hydrogen brittleness. The temperature of the occurrence of hydrogen brittleness decreases with increasing hydrogen content. Hydrogen brittleness of alloy VI15 occurs only at low rates of deformation in a narrow temperature interval from -30 to 100. The brittleness of $(\alpha + \beta)$ -titanium alloys depends upon processes occurring in the β -phase during Card

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EWI(1)/EWI(m)/EWP(t)/ETI IJP(c) ACC NRI AP6019642 SOURCE CODE: UR/0149/86/000/003/0094/0102 AUTHOR: Kolachev, B. A.; Livanov, V. A.; Bukhanova, A. A. ORG: Department of Metallography and Thermal Processing, Moscow Aviation Technologi cal Institute (Moskovskiy aviatsjonnyy tekhnologicheskiy institut. Kafedra metallove-TITLE: Phase diagram of the system titanium-oxygen-hydrogen $\nu 7$ SOURCE: IVUZ. Tsvetnaya metallurgiya, no. 3, 1966, 94-102 TOPIC TAGS: titanium compound, oxygen compound, hydrogen compound, phase diagram ABSTRACT: The isotherms of the equilibrium pressure of hydrogen in the system Ti-O-H were plotted at temperatures of 700 and 800C. Oxygen was found to increase the equilibrium pressure of hydrogen in the system, especially at a content of more than 5 wt. %. The isotherms have sharp bends corresponding to the transition from one phase region to another which permits finding the boundaries of all phase regions of the system in the investigated concentration range of oxygen and hydrogen except the interface between the α + β - and β regions. Isobars of the equilibrium pressure of hydrogen in the system were plotted at 700 and 800C, from which the position of the conodes in the two-phase region and the boundary between the $\alpha + \beta$ - and β -regions were established. The isothermal cross sections of the Card 1/2 UDC: 620.181.663.295'546.21'546.11

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LICERATY ENT(m)/EMP(w)/EMP(t)/ETI LIF(c) JD/JH ACC NR: AP6029676 AUTHORS: Kelachov, B. A.; Livanov, V. A.; Drozdov, P. D.; Bukhanova, A. A. ORG: none TITLE: Machanical proportion of alloy MA2-1 containing different concentrations of hydrogen SOURCE: Touchayo metally, no. 8, 1966, 88-90 TOPIC TAGS: magnesium alloy, hydrogen, hydrogen embrittlement / MA2-1 magnesium allow of its hydrogen content. The investigation was initiated to corroborate a mechanism A. A. Bukhanova, and N. Ya. Gusel'nikov (Novyye issledovaniya titanovykh splavov. Izd. Nauka. 1965 s. 212). The mechanical proporties of the alloy was initiated to corroborate a mechanism A. A. Bukhanova, and N. Ya. Gusel'nikov (Novyye issledovaniya titanovykh splavov.	Y
Izd. Nauka. 1965 s. 212). The mechanical properties of the specimens were ascertaine the specimens, determined after A. P. Gudchenko and A. K. Leont'yev (Sb. Trudy MATI, 1961, vyp. 49, s. 137), was 18 cm ³ and 9 cm ³ per 100 g respectively. The experimenta with the proposed dislocation hypothesis of hydrogen embrittlement.	
Card 1/2 UDC: 669.715:620,1	



ACC NR. AT'6036415 (A). SOURCE CODE: UR/2536/66/000/066/0063/0075 (Candidate of technical sciences); Shevchenko, V. V. (Engineer) ORG: none TITLE: Growth kinetics of β -grain in industrial titanium alloys SOURCE: Moscow. Aviatsionnyy tekhnologicheskiy institut. Trudy, no. 66, 1966. Struktura i svoystva aviatsionnykh staley i splavov (Structure and properties of aircraft steels and METAL PRESSING, TOPIC TAGS: Atitanium alloy, grain growth, phase composition, metalworking / VT5 Ti alloy, VT5-l Ti alloy, VT3-l Ti alloy, VT-6 Ti alloy, VT8 Ti alloy ABSTRACT: The kinetics of grain growth in the β -region determines the optimal temperature and heating time of press worked titanium alloys and hence it is an important consideration. When pressworked at temperatures corresponding to the \beta-region these alloys are more easy to process technologically and are more readily forged, since the β-phase with its cubic structure displays greater plasticity than the α-phase and there is no marked anisotropy of properties as is the case for forging at temperatures corresponding to the $(\alpha + \beta)$ region. So far, UDC: 669.017:669.295

however, Ti alloys have not been pressworked at temperatures corresponding to the \beta-region owing to the decrease in their plasticity, which is attributed to the intense growth of grain in this region. In this connection, the authors investigated the kinetics of grain growth in the β -region for the Ti alloys VT5, VT5-1 (α -alloys) and VT3-1, VT-6, VT8 (α + β alloys) at 1000-1200°C for from 0.5 to 12 hr. Findings: β-grain at first grows at a fairly intense rate; as the annealing continues, however, this rate slows down. The growth of β-grain obeys a parabolic law of $D = kt^n$ (where D = mean grain diameter; k = constant dependent on temperature and material; t - annealing time; n - exponent dependent primarily on the material and, to a smaller degree, on temperature). The dependence of grain size on temperature for the investigated α and $(\alpha + \beta)$ alloys may be separated into three temperature intervals within which grain grows differently: at 1000-1050°C the grain growth rate is insignificant, owing to the impeding effect of the impurities segregating at the grain boundaries; above 1050°C the effect of impurities disappears and the grain growth rate gets intensified; beginning with roughly 1100°C the grain growth rate decreases, probably because then the grains acquire more or less stable shape and size. Orig. art. has: 7 figures, 2 tables, 2 formulas. SUB CODE: 13, 11

SOB CODE: 13, 11 / SUBM DATE: none/ ORIG REF: 002

Card 2/2

SOURCE CODE: UR/2536/66/000/066/0076/0086

AUTHOR: Kolachev. B. A. (Candidate of technical sciences); Bukhanova, A. A. (Candidate of technical sciences); Livanov, V.A. (Doctor of technical sciences; Professor) ORG: none

TITLE: Phase distribution of hydrogen in $(\alpha + \beta)$ titanium alloys

SOURCE: Moscow. Aviatsionnyy tekhnologicheskiy institut. Trudy, no. 66, 1966. Struktura i svoystva aviatsionnykh staley i splavov (Structure and properties of aircraft steels and alloys),

TOPIC TAGS: Attanium alloy, hydrogen, phase composition / VT6 titanium alloy

ABSTRACT: While it is known (Livanov, V. A., Bukhanova, A. A., Kolachev, B. A. Vodorod v titane, Metallurgizdat. 1962) that hydrogen in $(\alpha + \beta)$ Ti alloys concentrates in the β-phase, the temperature dependence of the phase distribution of hydrogen still has not been established. Accordingly, the authors investigated the interaction between hydrogen and α - and β -phases of $(\alpha + \beta)$ Ti alloys (the alloys Ti + 0.5% Mo and Ti + 12.5% Mo, representing the α - and β -phases of the Ti-Mo system in an equilibrium at 800°C, and the alloys Ti +8% Al +2% V and Ti +4% Al +6% V, representing the α - and β -phases of the industrial alloy VT6

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UDC: 669.017:669.295

(Ti +6% Al +4% V) in an equilibrium at 900-925°C) at various temperatures. Hydrogen was added in small portions to the vacuum-annealed alloy specimens, each succeeding portion being introduced only after the equilibrium pressure due to the addition of the preceding portion had set in. The amount of the hydrogen absorbed by the specimen was determined according to the pressure difference in the system. Findings: in the two-phase region hydrogen is nonuniformly distributed between the phases. The ratio between hydrogen concentrations in the β - and α -phases is determined by the entropic factors and by the heats of dissolution of hydrogen in the phases. At low temperatures in $(\alpha + \beta)$ Ti alloys hydrogen gets concentrated in the β -phase, since the thermal effect of the dissolution of hydrogen in this phase (taking polarity into account) is smaller than in the α-phase. At temperatures of the order of 800-900°C hydrogen satisfactorily dissolves in the β- and α-phases; the hydrogen concentration in the β -phase of the Ti-Mo and Ti-Al-V systems is only 1.3-1.4 times as high as in the α -phase. As the temperature decreases hydrogen migrates from the α-phase to the β-phase until the hydrogen concentration ratio between these phases C_{β}/C_{α} increase to several tens. The thermodynamic analysis of the solutions of hydrogen in α - and β -phases of Ti alloys applies to any two--phase system. Thus, hydrogen concentrations in the α - and β -phases of $(\alpha + \beta)$ Ti alloys are similar at high temperatures, and it is only at sufficiently low temperatures that hydrogen concentrates in the β -phase. It follows hence that the proneness of $(\alpha + \beta)$ Ti alloys to hydrogen brittleness must markedly depend on the test temperature and on the previous heat treat-

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ACC NR AT6036418

SOURCE CODE: UR/2536/66/000/066/0096/0102

AUTHOR: Kolachev, B. A. (Candidate of technical sciences); Livanov, V. A. (Doctor of technical sciences, Professor); Brkhanova, A. A. (Candidate of technical sciences);

ORG: nono

TITLE: On the abrupt decrease in the plasticity of titanium at high temperatures

SOURCE: Moscow. Aviatsionnyy tekhnologicheskiy institut. Trudy, no. 66, 1966. Struktura i svoystva aviatsionnykh staley i splavov (Structure and properties of aircraft steels and alloys)

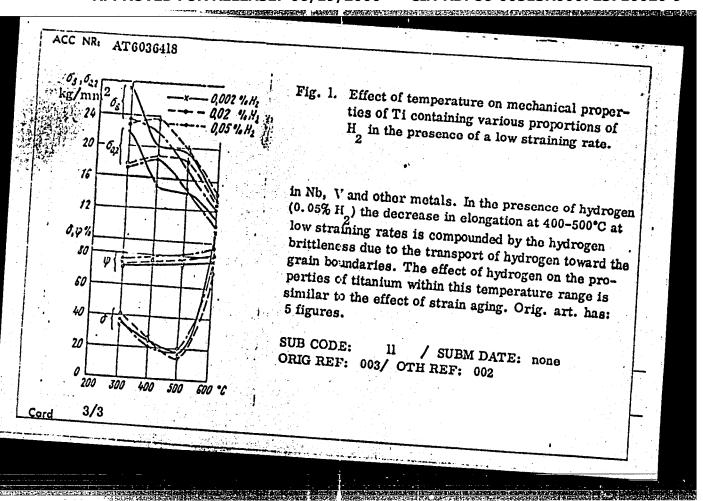
TOPIC TAGS: titanium, hydrogen, plasticity, brittleness, strain

ABSTRACT: According to a previous hypothesis by the first three of the authors (B. A. Kolachev et al. Issledovaniye titana i yego splavov, Izd-vo AN SSSR, 1963) the reason for the hydrogen brittleness of a number of metals is that the hydrogen-atom atmospheres forming at the dislocations are entrained by the latter in the presence of low straining rates, so that the hydrogen concentration at the grain boundaries or at other obstacles at which the disloca-

UDC: 669.017:669.295

ACC NR: AT6036418

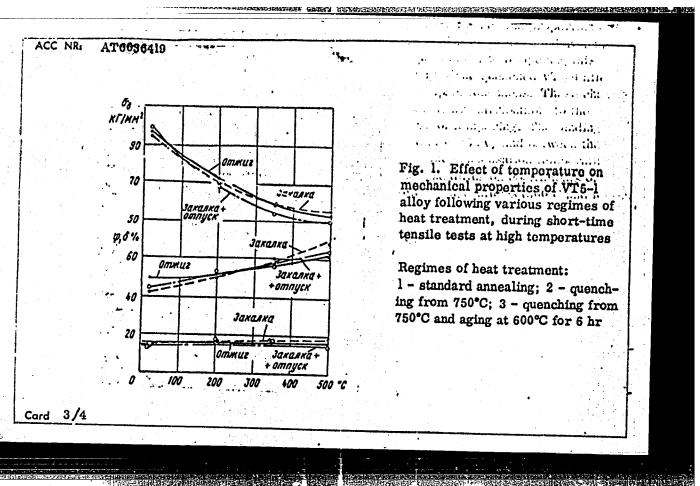
tions pile up becomes sufficient for a sharp acceleration of the development and propagation of cracks leading to fracture of the metal. Now the authors show that the hydrogen brittleness developing in the presence of low straining rates manifests itself within a temperature range (300-550°C) which corresponds to a specific value (10⁻⁵-10⁻⁶ cm²/sec) of the diffusion coefficient of hydrogen. In this connection, the authors investigate the effect of hydrogen on the mechanical properties of regular (0.002% H₂) and vacuum-annealed (0.02 and 0.05% H₂) rods of technically pure Ti subjected to tensile strength tests at normal (4 mm/min) and low (0. 4 mm/ /min) straining rates. Findings: the minimum elongation per unit length for Ti in the presence of normal straining rate was recorded at 500°C (Fig. 1) while in the presence of the below-nor normal straining rate (0.4 min/min) the mechanical properties of the Ti with 0.002% H₂ increase up to a point with increasing temperature whereas those of the Ti with 0.0% ${
m H}_2$ steadily decrease with increasing temperature. These experiments were organized on the assumption that the sharp decrease in the plasticity of Ti at high temperatures is due to hydrogen alone. The experiments revelaed, however, that this sharp decrease in plasticity within the temperature range of 300-550°C also occurs in technically pure Ti (0,002% H₂) — not as distinctly as in the Ti containing 0.05% H₂ but still distinctly enough. This sharp decrease is apparently due to the presence of 0_2 and N_2 and resembles similar phenomena discovered



ACC NR AT6036419 (A)SOURCE CODE: UR/2536/66/000/066/0103/0113 AUTHOR: Kolachev, B. A. (Candidate of technical sciences); Faynbron, S. M. (Engineer) ORG: none TITLE: Effect of quenching from the a-region and subsequent tempering on the mechanical properties of the titanium alloy VT5-1 SOURCE: Moscow. Aviatsionnyy tekhnologicheskiy institut. Trudy, no. 66, 1966. Struktura i svoystva aviatsionnykh staley i splavov (Structure and properties of aircraft steels and al-TOPIC TAGS: titanium alloy, metal heat treatment, tempering, mechanical property/ ABSTRACT: Contrary to the traditional theories, it is now known that quenching and tempering may induce structural changes in pure metals and single-phase alloys. These changes may be associated with Cottrell atmospheres, redistribution of dislocations and vacancies, interaction between defects and impurities, etc. In this connection, the authors attempted to utilize some of these effects to alter the properties of the α-Ti alloy VT5-1 (4.5% Al, 2.1% Sn, 0.04% UDC: 669.017:669.295

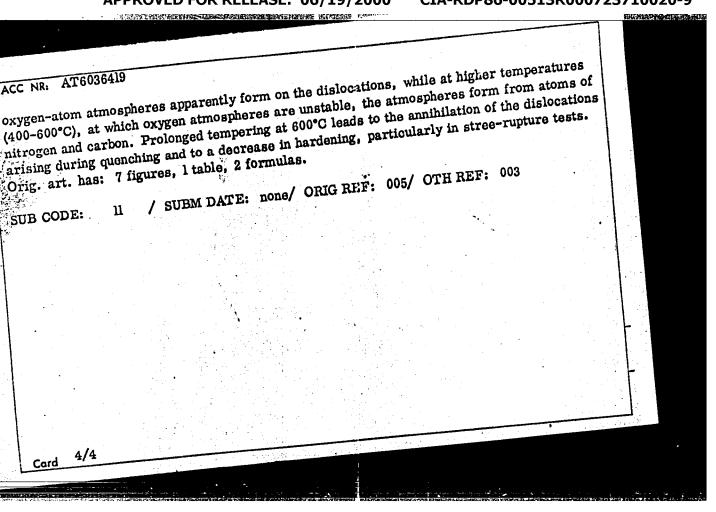
ACC NR: AT6036419

 N_2 , 0.15% 0_2 , 0.10% Fe, 0.15% Si, 0.003% H_2 , 0.05% C, with Ti as the remainder), whose temperature of transition from the α -region to the ($\alpha + \beta$) region is 950-975°C. Specimens of these alloys were quenched in water from 750, 850 and 950°C and tempered at 300, 400, 500, 600, and 700°C for from 1 to 7 hr. Findings: at room temperature the highest short-time tensile strength is displayed by specimens tempered at 600°C for 6 hr and the lowest, by specimens quenched from 750°C (Fig. 1). At higher temperatures the specimens display somewhat higher strength in quenched state than following standard annealing or tempering. A similar pattern is observed for the yield point and elongation. Thus, by means of quenching from the α -region and subsequent tempering it is possible to markedly influence the properties of the VT5-1 alloy and thus sometimes even correct defective castings: following its standard annealing this alloy has an ultimate strength of 96.9 kg/mm²; by quenching and tempering this strength can be adjusted to from 89 to 98.3 kg/rnm². Tempering of the quenched VT5-1 alloy at 300-600°C enhances its strength properties and reduces its impact toughness. These changes in properties following quenching from the α -region and tempering are attributable to the formation of impurity atmospheres on dislocations in the course of tempering. The binding energy between the oxygen atoms and dislocations is estimated at 0.25 ev, and between the atoms of nitrogen and carbon 144 approximately 0.4 ev, so that the condensation temperatures of impurity atmospheres miles by roughly 300 and 500°C. Since the diffusion coefficients of oxygen and titanium are little than those of nitrogen, at lower temperatures (300°C and lower)



"APPROVED FOR RELEASE: 06/19/2000

CIA-RDP86-00513R000723710020-9



ACC NR: AT6036413

SOURCE CODE: UR/2536/66/000/006/0039/0052

AUTHOR: Kolachev, B. A. (Candidate of technical sciences); Livanov, V. A. (Doctor of technical sciences, Professor); Vishnyakov, D. Y. (Doctor of technical sciences, Professor); Lyasotskaya, V. S. (Engineer)

ORG: none

TITLE: Isothermal transformations in alloys of titanium with molybdenum

SOURCE: Moscow. Aviatsionnyy tekhnologicheskiy institut. Trudy, no. 66, 1966. Struktura i svoystva aviatsionnykh staley i splavov (Structure and properties of aircraft steels and alloys), 39-52

TOPIC TAGS: isothermal transformation, titanium base alloy, molybdenum, phase diagram, martensitic transformation

ABSTRACT: The literature on the isothermal transformations of alloys in the Ti-Mo system shows certain gaps. Thus, e.g. Bungardt and Ruedinger (Z. Metallkunde, 1961, no. 52(2)) specify below the initial temperature M_i of martensitic transformation only the line of the beginning and end of decomposition of the α' -phase whereas both the β -phase and the α' -phase

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UDC: 669,017:669.295'28

should isothermally decompose within the temperature range between M_i and the final temperature M_f of martensitic transformation. To fill this gap the authors investigated specimens of titanium alloys containing 2, 6, 9 and 13% Mo and, on the basis of the change in hardness following isothermal treatment and according to the results of metallographic, selective radiographic and dilatometric analyses, they constructed the pertinent isothermal transformation diagrams. Isothermal treatment of the specimens was accomplished by placing them in an electric furnace at 1000°C for 1 hr and thereupon transferring them to tin, lead or salt baths (at 300, 400 and 500-800°C, respectively) and, after definite intervals of time, cooling them in water. Findings: the isothermal transformation diagram (ITD) for the alloy Ti+2% Mo is represented by two series of lines describing the beginning and end of the decomposition of the β - and α' -phases. Within the temperature range from M_i to M_f these two series of lines overlap; the same applies to the ITD for the alloy Ti+6% Mo. On the other hand, the ITD for the alloy Ti+6% Mo also includes a line of formation of the w-phase (at temperatures of < 450°C). For the alloy Ti+13% Mo the ITD is represented by lines of the beginning and end of decomposition of the β -phase and by a line restricting the region of existence of the w-phase. These lines overlap and the region $(\alpha+\beta+\omega)$ appears on the diagram. Thus increasing the Mo content above 9% complicates the formation of the w-phase and shifts to the right the lines of the beginning of the segregation of this phase. The isothermal decomposition of the o'-phase in Ti alloys is usually accompanied

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ACC NR: AT6036414

SOURCE CODE: UR/2536/66/000/066/0053/0062

AUTHOR: Vishnyakov, D. Ya. (Doctor of technical sciences, Professor): Kolachev, B. A. (Candidate of technical sciences); Lyasotskaya, V. S. (Engineer); Lebedeva, V. D. (Engineer)

ORG: none

TITLE: Isothermal transformations in alloys of titanium with chromium

SOURCE: Moscow. Aviatsionnyy tekhnologicheskiy institut. Trudy, no. 66, 1966. Struktura i svoystva aviatsionnykh staley i splavov (Structure and properties of aircraft steels and alloys) 53-62

TOPIC TAGS: titanium base alloy, chromium, isothermal transformation, phase diagram

ABSTRACT: The literature on this subject so far provides no information on isothermal transformations in alloys of the Ti-Cr system with hypo- and hypercutectoid compositions. To fill this gap, the authors constructed isothermal transformation diagrams (ITD) in alloys of Ti with 6 and 11% Cr (hypocutectoid), 15% Cr (cutectoid) and 20% Cr (hypercutectoid) according to the change in hardness with isothermal treatment as well as according to the results of metallographic, radiographic and dilatometric analyses. Isothermal treatment at 600°C was

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UDC: 669.017:669.295126

ACC NR: AT6036414

accomplished by rapidly cooling the specimens from a high temperature to the temperature of treatment, and at 550°C and below, after quenching. In both cases the isothermal treatment at > 300°C was performed in lead baths, and at 300-100°C, in baths of Wood's alloy. Findings: the hypocutectoid and hypereutectoid alloys display two minima of β -phase stability: the low-temperature minimum, associated with the formation of the ω -phase, and the high-temperature minimum, conditioned by the hypocutectoid segregation of the α -phase or TiCr₂. Increasing the Cr content above 6% complicates the segregation of the ω -phase and shifts to the right and downward the lives of the commencement of this correspicant. The critical formation of the ω -phase decomposes nonuniformly; this is due not so much to the chemical heterogeneity of grains as to the heterogeneity of substructure, arising on rapid cooling of specimens or during the subsequent isothermal treatment. This substructure forms as a result of thermal stresses and the subsequent redistribution of dislocations. Orig. art. has: 10 figures.

SUB CODE: EL 11, 20/ SUBM DATE: none/ORIG REF: 003/ OTH REF: 006

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ACC_NR, AT 0036417

SOURCE CODE: UR/2536/66/000/066/0087/0095

AUTHOR: Kolachev. B. A. (Candidate of technical sciences); Lyasotskaya, V. S. (Engineer)

ORG: none

TITLE: Effect of hydrogen on the processes occurring in the VT3-1 titanium alloy during aging

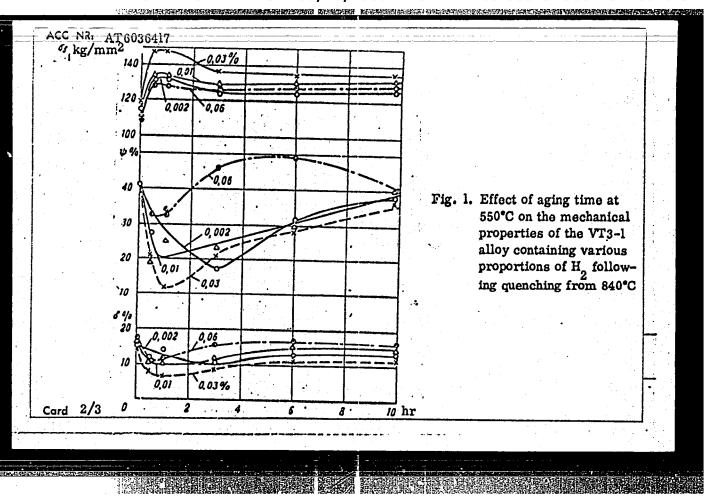
SOURCE: Moscow. Aviatsionnyy tekhnologicheskiy institut. Trudy, no. 66, 1966. Struktura i svoystva aviatsionnykh staley i splavov (Structure and properties of aircraft steels and alloys), 87-95

TOPIC TAGS: titanium alloy, hydrogen, metal aging, phase composition

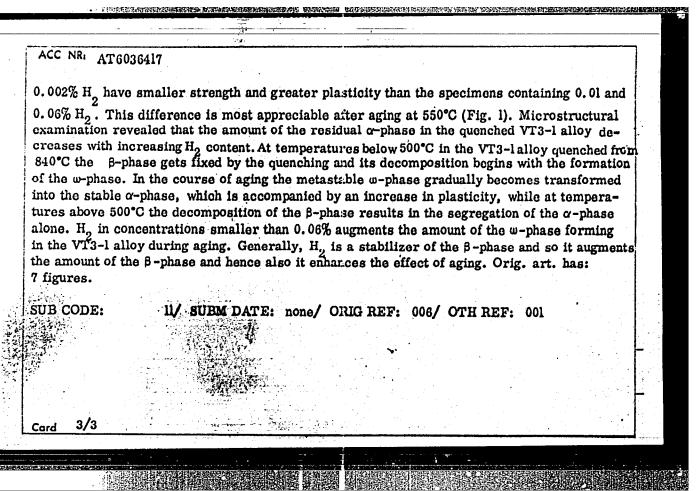
ABSTRACT: Specimens of the VT3-1 titanium alloy (5.5% Al, 1.91% Cr, 2.05% Mo, 0.23% Si, 0.2% Fe, 0.04% C, 0.11% O_2 , 0.01% H_2 , with Ti as the remainder) were heated in electric furnaces at 840°C for 1 hr and water-quenched, after which they were aged for 2, 4, 6, 8 and 10 hr at 400, 500, 550 and 600°C, with subsequent water quenching. The H_2 content of some of the specimens was increased to 0.03 and 0.06%, and of others, reduced by vacuum annealing to 0.002%. Mechanical tests of the specimens showed that the specimens containing

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UDC: 669.017:669.295



APPROVED FOR RELEASE: 06/19/2000 CIA-RDP86-00513R000723710020-9"



ACC NR: AMG026330

Monograph

UR/

Kolachev, Boris Aleksandrovich...

Hydrogen brittleness of nonferrous metals (Vodorodnaya khrupkost' tsvetnykh metallov) Moscow, Izd-vo "Metallurgiya", 1966. 255 p. illus., biblic. 3100 copies printed.

TOPIC TAGS: motalines, metal brittleness, hydrogen induced brittleness, mutal, hiddeness, hydrogen, hiddeness, hydrogen induced brittleness, mutal, hiddeness, hydrogen, hiddeness, hydrogen allows and engineers as well as by students doing advanced or graduate work. The book is devoted to problems concerning the interaction of hydrogen with metals, and the harmful effects this has on the properties of the metal. Considerable attention is given to processes that take place during the hydrogen-metal interaction, the state of hydrogen in liquid and solid solution, and the interaction of hydrogen with dislocations and other structural imperfections of metals. The effect of hydrogen on the structure and properties of the following metals are discussed: Be, Mg, Al, U, Ti, Zr, V, Nb, Ta, Cr, Mo. W. Pt, Cu, Ag, Au. Special attention is devoted to the hydrogen brittleness of Ti alloys.

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UDC: 669.2/.8:539.56

ACC NR: AM6066330 Part One. Interaction of metals with hydrogen and specific structural features in metal-hydrogen systems 2. Ch. 1. General patterns in metal hydrogen interaction -- 5 3. Ch. 2. State of hydrogen in metals -- 18 Part Two. General problems concerning the effects of hydrogen on the mechanical properties of metals 4. Ch. 1. Classification of types of hydrogen-brittleness in metals -- 44 5. Ch. 2. Hydrogen brittleness of the first type -- 48 6. Ch. 3. Hydrogen brittleness of the second type -- 63 7. Ch. 4. Effect of hydrogen on the creep, fatigue resistance, and weldability -- 122 Part Three. Effect of hydrogen on the structure and properties of metals and alloys 8. Ch. 1. Beryllium, magnesium, and aluminum -- 129 9. Ch. 2. Uranium -- 151 10. Ch. 3. Titanium and its alloys -- 157 11. Ch. 4. Zirconium and its alloys -- 192
12. Ch. 5. Vanadium, niobium, and tantalum -- 203 13. Ch. 6. Chronium, molybdemum, and tungsten -- 220 14. Ch. 7. Nickel, palladium, and platinum -- 226 15. Ch. 8. Copper, silver, and gold -- 233 16. Bibliography -- 239 SUPM DATE: 02Jan66/ ORIG REF: 143/ OTH REF: 348/ SUB CODE: 11/ **Card** 2/2

ACC NR. AP7002867 BOURCE CODE: UR/0149/66/000/006/0142/0145 AUTHOR: Kolachev, B.A.; Livanov, V.A.; Bukhanova, A.A.; Gusel'nikov. N. Ya.; Lyasotskaya, V.S. ORG: Department of Metal Science and Technology of Thermal Processing of Ketals, Moscow Aviation Technology Institute (Moskoyskiy aviatsionnyy tekhnologicheskiy institut, kaiedra metalloyedeniya i tekhnologii termicheskoy obrabotki metalloy tekhnologii termicheskoy obrabotki metalloy ariously heat-treated VT3-1 alloy IVUZ. Tsvetnaya metallurgiya, no. 6, 1966, 142-145 SOURCE TOPIC TAGS: Atitanium alloy, transcript route in the transcript alloy, alloy strength, willow brittlemess, alloy structure/VT3-1 alloy Anthry tiluctility. ABSTRACT: Hydrogen-induced embrittlement of VT3-1 and other a + B titanium alloys depends not only on the hydrogen content, but to a considerable extent on the content of other impurities, heat treatment, grain size and the type and conditions of deformation. To determine the effect of the various factors, several series of specimens of modified (with increased Al, Fe and Si content) VT3-1 [U.S. Ti 155A] titanium alloy with a hydrogen content of up to 0.1 wt.% were annealed at 800C and slowly cooled, or annealed at 840 or 970C, quenched, aged at 550C for 0.5-3 hr and then 1/2 UDC: 669.018.1

ACC NR: AP7002867

subjected to tension tests at a deformation rate of 0.4-4.0 mm/min. In the alloy annealed and slowly cooled, a hydrogen content of up to 0.1% had no significant effect on the ductility of the alloy at a deformation rate as low as 0.4 mm/min, while in the as-quenched alloy deformed at the same rate, a substantial decrease in the reduction of area occurred at a hydrogen content of 0.003% H. However, at a strain rate of 4 mm/min, no noticeable change in the reduction of area was observed in as-quenched alloys containing up to 0.05% H2. The brittleness of as-quenched alloy increased with increasing annealing temperature, since this decreased the amount of residual a-phase and increased the amount of the a-phase. An especially strong effect of hydrogen was observed in aged VT3-1 alloy. Short (0.5 hr) aging at 550C significantly increased the tensile and yield strengths of the alloy containing 0.03 and 0.05% hydrogen and sharply decreased the elongation and reduction of area. The allow strength decreased and ductility increased with increasing aging time from 0.5 to 3 hr, but changed only slightly with still longer aging. The second of the second of the second second of the secon

SUB CODE: 11, 13/ SUBM DATE: 27Apr65/ ORIG REF: 005/ OTH REF: 001
ATD PRESS: 5114

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[MS]

KOLACHEV M. P.

USSR / General and Special Zoology. Insects: Harmful Insects and Mites. Fruits and Berry Crop Pests.

Abs Jour: Ref Zhur-Biol., No 1, 1959, 2335.

Author : Kolachev. M. P.

: Turkmenien Institute of Agriculture. Inst

! The Harmfulness of the Big Cotton-Boll Cicada Title (Cicadatra ochreata Mel.) in Turkmen Gardens.

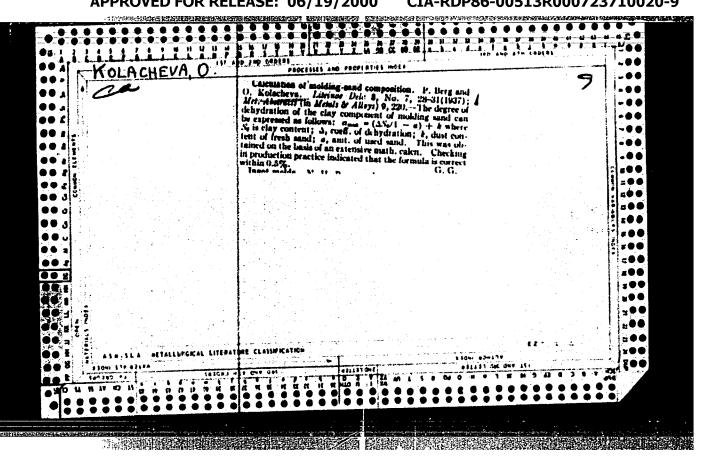
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Orig Pub: Tr. Turkme s.-kh. in-ta, 1957, 9, 83-85.

Abstract: The cicada (C) damages one-year shoots, while laying its eggs, most often on the sunlight side. The C makes 10.5 pricks per deposit in large-scale egg-deposits on an apple tree along 11.8 cm of the shoot, in small deposits it makes 4.8 pricks per 3.8 cm of the shoot, on the cherry

tree it makes 14.7 pricks per 13 cm, on the

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SOV/137-57-1-783

Translation from: Referativnyy zhurnal. Metallurgiya, 1957, Nr 1, p 100 (USSR)

AUTHOR:

Kolacheva, O. V.

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TITLE:

Some Problems Arising in the Production of Castings in Shell Molds Made With a Bakelite Binder (Nekotoryye voprosy polucheniya otlivok v obolochkovykh formakh na bakelitovom svyazuyushchem)

V sb.: Novoye v teorii i praktike liteyn. proiz-va. Moscow-Leningrad, Mashgiz, 1956, pp 373-381 PERIODICAL:

ABSTRACT: Powdered bakelite and bakelite varnish (BV) were used as binders for the preparation of shell molds (SM). Molding mixtures (MM) to be made of powdered bakelite were prepared by the usual method; MM with BV were prepared in a pug mill after preheating to 50°C; the mixtures were then dried, milled, and screened through a Nr-40 screen. The utilization of a small amount of BV along with the powdered bakelite decreases the dust emission of MM and its sticking to the metal pattern and also promotes the uniform distribution of the binder among the sand grains. The ob in tension of MM, determined on figure eights 10 mm thick, when the mixture contains 6% powdered bakelite and BV is 30 - 35 and 20 - 25 kg/cm²,

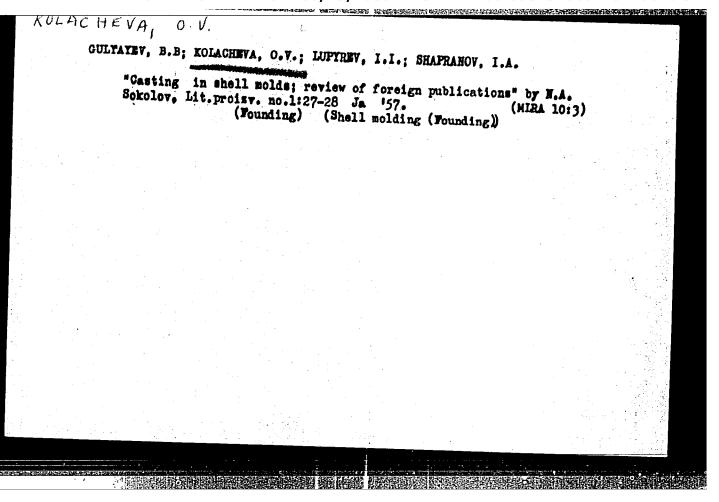
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Some Problems Arising in the Production of Castings in Shell Molds (cont.)

respectively (for K 70/140 sand); the use of finer sand greatly decreases the σ_b in tension of SM (to 7 kg/cm² for 270-undersize sand with 10% binder). For the preparation of SM the author recommends the use of refractory materials with the particle-size modulus in the 70-200 range. For steel casting the use of magnesite, chrome-magnesite, chromite ore, etc., is recommended. For the determination of the rate of accumulation of MM on the hot pattern the following relationship was established: $X = k\sqrt{\tau}$, where X is the thickness of the SM (in mm), τ is the time the MM remains on the plate (in sec), k is a coefficient; when the patterns used are heated to 150, 180, and 250°; k equals 1.73, 2.05, and 2.45 mm/sec $^{1/2}$, for k which were determined under the same conditions for sand, chrome-magnesite, material is returned to the hopper, the sand SM collapse easier than the excess magnesite or chromite SM. When SM are prepared on flat patterns < 10 mm high pouring of MM into a frame 6-7 mm higher than the pattern is recommended to

Ya. M.

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KOLACHEVA, O.V.

25(1) ^b

PHASE I BOOK EXPLOITATION SOV/1440

Nauchno-tekhnicheskoye obshchestvo mashinostroitel'noy promyshlennosti. Leningradskoye oblastnoye pravleniye

Lit'ye povyshennoy tochnosti (High-precision Casting) Moscow, Mashgiz, 1958. 196 p. (Series: Its: Sbornik, kn.45) 7,000 ccpies printed.

Ed.: A.N. Sokolov; Tech. Ed.: L.V. Sokolova; Managing Ed. for Literature on Machine-building Technology (Leningrad Division, Mashgiz): Ye. P. Naurov, Engineer.

PURPOSE: This book is intended for engineers and technicians at foundries and planning and research institutes.

COVERAGE: The book contains the transactions of a special conference called in November, 1956, by the Leningrad Oblast Administration of the Nauchno-tekhnicheskoye obshchestvo NTO (Scientific and Technical Society of the Machine-building Industry). The articles describe advanced techniques used in

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High-precision Casting

SOV/1440

precision-casting processes such as shell molding, investment casting, pressure die casting, press die casting (called in Russian "forging of liquid metal"), and suction casting. Special attention is given to the production of large precision castings, one of the principal problems in the industry. At the same time, methods of improving the precision of sand-mold castings are examined. Experience gained in the mechanization of precision-casting and shell-molding processes is reported. Information is given on the present state of precision casting, both in the USSR and elsewhere. No personalities are mentioned.

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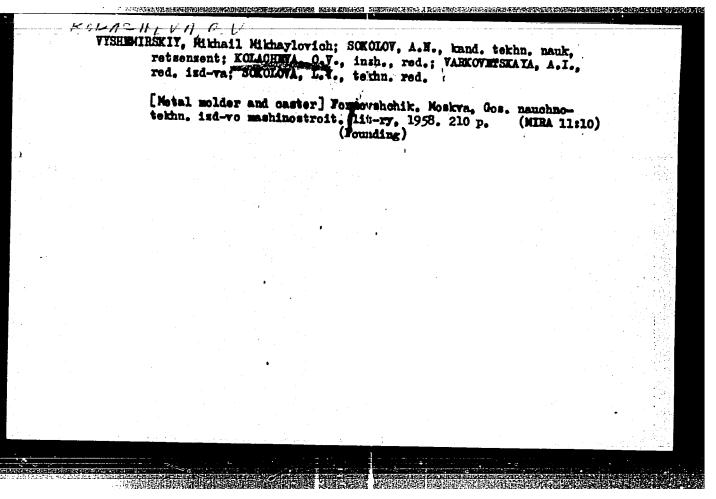
Shub, I. Ye. [Chairman, Committee on Special Methods of Casting, Leningrad Oblast Administration of the Scientific

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APPROVED FOR RELEASE: 06/19/2000 CIA-RI

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and Technical Society of the Machine-building Industry. Equipment for Producing Castings in Shell Molds	18
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S/123/60/000/021/004/004 A005/A001

Translation from: Referativnyy zhurnal, Mashinostroyeniye, 1960, No. 21, p. 197, # 116748

AUTHOR:

Kolacheva, O. V.

TITLE:

Substitutes of Ethyl Ortho-Silicate for Investment Casting

FERIODICAL: Sb. statey po formovochn. materialam. Moscow, 1958, pp. 40-49

TEXT: Two kinds of substitutes of ethyl ortho-silicate were developed and introduced into practice: the binder AC (DS) being the product of treatment of a water glass solution by electrodialysis, and the binder KC(KS), being the product of treatment of a water glass solution by ion exchange. The author recommends to use, for the treatment of water glass solutions by electrodialysis and ion exchange, hydrochloric acid solutions the composition of which is calculated according to the chemical analysis of the initial water glass. The content of silicon dioxide in the hydrochloric acid solution subjected to the treatment by electrodialysis has to be not greater than 7.5%, and for the treatment by ion exchange not greater than 10%. The quantity of water glass, hydrochloric acid, and distilled water for the preparation of the hydrochloric acid solution must be

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S/123/60/000/021/004/004 A005/A001

Substitutes of Ethyl Ortho-Silicate for Investment Casting

taken with the consideration of the mentioned limiting contents of silicon dioxide The solutions of water glass and hydrochloric acid are prepared separately in clean vessels; 68% of the distilled water volume is used for the dilution of the water glass, and 32% for the dilution of the hydrochloric acid. The vessel with the diluted hydrochloric acid. The vessel with the diluted hydrochloric acid is set under the paddle mixer, and the mixed water glass solution is added carefully in a fine stream at continuous mixing. The reverse sequence of infusion is not admissible. The DS binder is obtained by electrodialysis of the hydrochloric acid solution in a wooden or rubber tank divided into 3 sections by semipermeable diaphragms (cellophane foils). The hydrochloric acid solution is filled into the middle section, the by-sections are charged with tap water. The electrodes are arranged at a distance of 100-120 mm in the by-sections. The cathode is a tin plate, the anode is a graphite-for a corbon rod. The electro-dialysis is performed until obtaining the solution viscosity of 1.7. The prepared solution is cooled down to room temperature and diluted with ethyl alcohol. The duration of storage is 3-5 days. The KS binder is obtained by filtration of the hydrochloric acid solution of water glass through a layer of cationite-resin absorbing the sodium

Card 2/3

SOV/123-59-15-60525

Translation from: Referativnyy zhurnal. Mashinostroyeniye, 1959, Nr 15, p 232 (USSR)

AUTHOR:

Kolacheva, O.V.

TITLE:

Investigation of the Thermal Conditions of Solidification of Castings in

Shell Molds

PERIODICAL:

V sb.: Zatverdevaniye metallov. Mossow. Mashgiz, 1958, pp 231 - 242

ABSTRACT:

As shell molds have thin walls peculiar thermal conditions of solidification and cooling of castings prevail. Tests for the investigation of the thermal conditions of shell molds made of a bakelite mixture and of the steel solidification process in them were carried out in casting a box-type mold measuring $100 \cdot 100 \cdot 210$ mm with the following thicknesses of walls: 5 mm, 15 mm, 30 mm. The molds were made of a K50/100 sand mixture with 6% of bakelite lacquer and were filled with steel of the 35L grade. The results of temperature measurements during the filling of castings of different thickness are stated. Based on the temperature curves temperature fields were plotted of the mold walls and castings at various time instants from the beginning of the filling. At the free pouring of the shell molds with a bakelite binder, a second heating source of the mold walls is created by the burning of the

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Investigation of the Thermal Conditions of Solidification of Castings in Shell Molds

bakelite which changes the character of the temperature field in the mold walls. When applying filling the temperature field in the mold is analogous to the pouring into sand molds. The analysis of the curves of solidification of castings with different thickness of walls, obtained by measuring temperatures, permitted to calculate the coefficient of solidification which was equal to 0.1 sec 1/2 and to plot the theoretical graph of solidiffication. The methods of tests were worked out and investigations were carried out of the strength of the bakelite mixture at high temperatures and of the resistance to heat of the shell molds under the pouring conditions investigated. The design of the pattern was improved upon and a test device manufactured. The specimen to be tested was clamped in the chuck of the device, then a furnace, heated up to 700°C, was moved towards it and the specimen was subjected to heat during 30 sec, and afterwards it was destroyed by stretching. Even at a short spell of high temperature the strength of the mixture is lowered considerably and amounts to 10 - 15% of the initial one only. Methods of determining the resistance to heat of the mold were developed on the basis of hydrostatics since actually the resistance of the mold is reduced immediately after being filled with metal while pressure affects the mold only in the already filled part of the casting. The minimum wall thickness of the mold necessary to preserve it until a hard

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Investigation of the Thermal Conditions of Solidification of Castings in Shell Molds

skin of metal is formed should be not less than 8.5 mm. The strength of the shell mold when filled with steel does not exceed 1.7 kg/cm², while the grain size of the mixture and the character of the thermoreactive binder do not affect the resistance to heat of shell molds. The duration of the pouring process has a considerable effect on the resistance to heat of the shell molds, therefore they should be filled rapidly and through a dispersed gate system. 8 figures.

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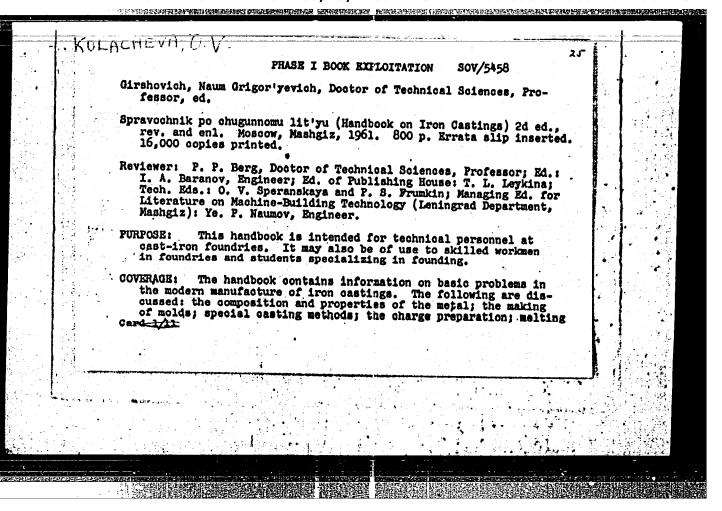
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GULYAYEV, Boris Borisovich. Prinimali uchastiye: SHAPRANOV, I.A., kand.tekhn. nauk; MAGNITSKIY, O.N., kand.tekhn.nauk; POSTNOV, L.M., kand.tekhn. nauk; BOROVSKIY, Yu.F., kand.tekhn.nauk; KOLACHEVA, O.V., kand. tekhn.nauk, BERG, P.O., prof., doktor tekhn.nauk; KolaCHEVA, o.V., kand. tekhn.nauk, BERG, P.O., prof., doktor tekhn.nauk; KolaCHEVA, o.V., kand. tekhn.nauk; KolaCHEVA, O.V., tekhn.nauk; KolaCHEVA, A.I., nauchnyy red.; CHFAS, M.A., red.izd-va; KONTOROVICH, A.I., tekhn.red.; SPERANSKAYA, O.V., tekhn.red.

[Founding processes] Literary protessey. Moskva, Gos.nauchno-tekhn.isd-vo mashinostroit.lit-ry, 1960. 415 p.

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Handbook on Iron Castings	80V/5458	7.	
and modifying the cast iron; pour of castings; heat-treatment metho jection of castings. Information the mechanization of castings pro authors thank Professor P. P. Ber and staff members of the Mosstank metallurgist G. I. Kletskin, Cand their assistance. References fol references, mostly Soviet.	ds; and the inspection and of the foundry equipment and or duction is also presented, by, Poctor of Technical Science of Technical Science of Technical Science of Technical Science	re- n The noes, hief	
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AUTHOR: Berlyand, O. S.; Yerokhina, R. A.; Kolacheva, Z. A.

ORG: none

TITLE: Exchange of air masses between the stratosphere and troposphere in the Northern Hemisphere

SOURCE: AN ESSR. Mezhduvedomstvennyy geofinicheskiy komitet. Geofizicheskiy byulleten', no. 17, 1966, 55-58

TOPIC TAGS: atmospheric circulation, stratosphere, troposphere, atmospheric temperature, temperature distribution

ABSTRACT: This article presents the results of an investigation of the mechanism of exchange of air masses between the troposphere and stratosphere for given mean annual zonal distributions of temperature in the 0—16 km layer and the distribution of atmospheric pressure on the Earth's surface by finding a wind velocity field for determining the vertical motion of air masses. It was calculated that during a year an air mass weighing 3·10¹t, which amounts to 5% of the weight of the entire atmosphere, descends from the tropospuse in the 25—35°N sone. The weight of the 10—16-km air layer amounted to approximately 1/6 of the weight of the entire atmosphere. Thus, it is concluded that exchange of the entire air mass between the troposphere and stratosphere occurs within about 3.5 years in the 25—35°N region. Orig. art.

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USSR/Physics - Magnetic flux

FD-1485

Card 1/1

: Pub. 146-8/20

Author

Grachev, A. A.; Goronina, K. A.; Kolachevskiy, N. N.; and Andrianova,

Title

: Experimental investigation of variation of magnetic flux in a cable at

reversal of magnetization of one domain

Periodical

: Zhur. eksp, i teor. fiz., 27, 313-317, Sep 1954

Abstract

: Results of experimental investigation of magnetic flux generated in a single domain of a ferromagnetic cable are outlined. Experimental data concur within 10% accuracy with theoretical computation by S. M. Rytov

(ibid, 307-312). Four references.

Institution : Physicotechnical Institute, Gor'kiy State University

Submitted

: December 28, 1953

TOLHCHEVSKIY ANTIQUE ANTIQU N.N. . 2 PHASE I BOOK EXPLOITATION Moscow. Fiziko-tekhnicheskiy institut 1127 Issledovaniya po fizike i radiotekhnike (Research in Physics and Radio Engineering) Moscow, Oborongis, 1958. 132 p. (Series: Its Trudy, Ed.: Zaytseva, K.Ya., Engineer; Ed. of Publishing House: Gortsuyeva, N.A.; Tech. Ed.: Rozhin, V.P.; Managing Ed.; Zaymovakaya, A.S., Engineer. PURPOSE: The book may be useful to scientific personnel, engineers, and students conducting research in physics and radio engineering. COVERAGE: The book is a collection of 13 articles written by instructors and graduate and undergraduate students of the Moscow Institute of Physics and Technology. The articles discuss problems in radio physics, optics and physics. No personalities are mentioned. References appear at the end of each article. TABLE OF CONTENTS: Kozel, S.M., Candidate of Physical and Mathematical Sciences. Modulation Optical Interferometer for Measuring the Angle of a Light Beam Card 1/3

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<u>CIA-RDP86-00513R00072371002</u>

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PHASE I BOOK EXPLOITATION

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Moscow. Fiziko-tekhnicheskiy institut

- Issledovaniya po fizike i radiotekhnike (Research in Physics and Radio Engineering) Moscow, Oborongiz, 1959. 170 p. (Series: Its: Trudy, vyp. 4) Errata slip inserted, 2,150 copies printed.
- Sponsoring Agency: RSFSR. Ministerstvo vysshego i srednego spetsial'nogo obrazovaniya.
- Ed.: K.Ya. Zaytseva, Engineer; Ed. of Publishing House: S.D. Antonova; Tech. Ed.: L.A. Garnukhina; Managing Ed.: A.S. Zamovskaya, Engineer.
- PURPOSE: This book is intended for scientific workers, students in advanced courses and engineers.
- COVERAGE: This is a collection of 15 studies dealing with problems of radio physics, electronics, quantum physics, and aerodynamics. The studies examine the method of least squares as applied to the propagation of radio waves in the presence of a plane junction, the general conditions of stability of a random process at the output of a linear filter while a periodic unstable Card 1/0

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random process is supplied at the input of the filter, the results of experiments with a ferromagnetic specimen with large Barkhausen jumps as an explanation of the noise mechanism in ferromagnets at cyclic magnetization reversal, experiments for the determination of thermal characteristics and the results of an experimental study of a turbulent boundary layer in a supersonic flow. No personalities are mentioned. References accompany most articles.

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Kozel, S.M. [Candidate of Physics and Mathematics]. Transformation of Periodically Unsteady Fluctuations by Means of a Linear Filter A general expression for the correlation function of fluctuations at the output of a linear filter while a periodically unsteady random process is being fed into its input is derived. Stability conditions

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for the process are determined. Exact values of Fourier expansion coefficients of the correlation function for an oscillation circuit with a high-quality factor used as a filter and a process at the input representing a modulated periodic white noise are established.

Kolachevskiy, N.W. Ferromagnetic Core With Large Barkhausen Jumps in an Alternating Magnetic Field

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Problems concerning the emergence of cyclic magnetization reversal noise in specimens with a single Barkhausen jump are discussed. The continuous spectrum of inf induction in an induction coil, taking into account the induction reversal component, is determined. The experiments with large Barkhausen jumps show that it is erroneous to explain noise by the temporary fluctuations of the jump emergence moments. Noise should be calculated on the basis of fluctuations of the magnetic moment at the jump. The inclusion of the reversal component results in a drop of the spectral noise density at frequencies of $\omega = 3\omega$ and conversion of the spectral intensity to zero at $\omega = 0$. In this connection it is pointed out that the extrapolation of the spectral intensity curve to a different value than zero at $\omega \rightarrow 0$ as it is done by

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